

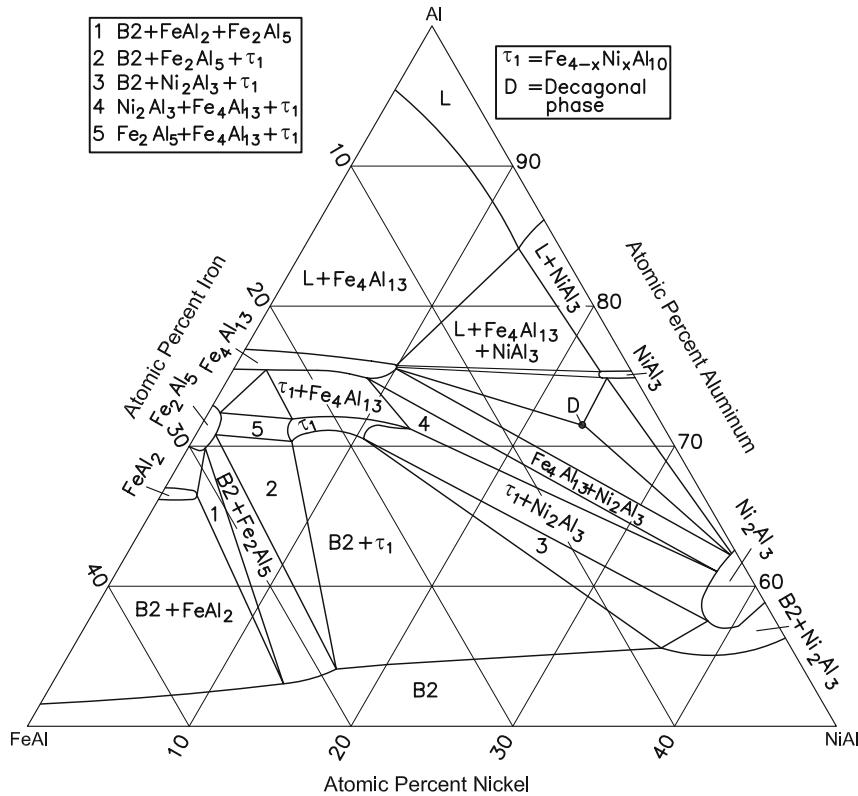
# Al-Fe-Ni (Aluminum-Iron-Nickel)

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In view of the technological importance of this ternary system in the development of high-temperature alloys, magnetic materials, and shape memory devices, it has been investigated experimentally a number of times. Many reviews and updates have also appeared in the last two decades [1988Ray, 1992Bud, 1994Rag, 1995Vil, 2005Cac, 2005Rag, 2006Ele, 2006Rag]. For a recent detailed summary, the reader is referred to the updated version of [1992Bud] by [2005Cac] and the review of [2006Ele]. [2006Ele] presented assessed experimental information on phase relationships and on thermochemical data, and Calphad-type/CVM calculations. Recently, [2007Chu] reinvestigated the Al-rich region and constructed an isothermal section at 850 °C, five vertical sections and a complete liquidus projection for this region. This update will be limited to the new results of [2007Chu].

## Binary Systems

The Al-Fe phase diagram [1993Kat] shows that the face-centered cubic (fcc) solid solution based on Fe is restricted by a  $\gamma$  loop. The body-centered cubic (bcc) solid solution exists in the disordered  $A2$  form ( $\alpha$ ), as well as the ordered  $B2$  and  $D0_3$  forms. Apart from the high temperature phase  $\varepsilon$ , there are three other intermediate phases in this system:  $FeAl_2$  (triclinic),  $Fe_2Al_5$  (70-73 at.% Al, orthorhombic), and  $FeAl_3$  or  $Fe_4Al_{13}$  (74.5-76.6 at.% Al, monoclinic). The Al-Ni phase diagram [1993Oka] shows five intermediate phases:  $NiAl_3$  ( $D0_{11}$ ,  $Fe_3C$ -type orthorhombic),  $Ni_2Al_3$  ( $D5_{13}$ -type hexagonal),  $NiAl$  ( $B2$ ,  $CsCl$ -type cubic, denoted  $\beta$ ),  $Ni_3Al_3$  ( $Ga_3Pt_5$ -type orthorhombic), and  $Ni_3Al$  ( $L1_2$ ,  $AuCu_3$ -type cubic, denoted  $\gamma'$ ). The Fe-Ni phase diagram [1993Swa] is characterized by a very narrow solidification range with a peritectic reaction at 1514 °C, between bcc



**Fig. 1** Al-Fe-Ni isothermal section for Al-rich alloys at 850 °C [2007Chu]

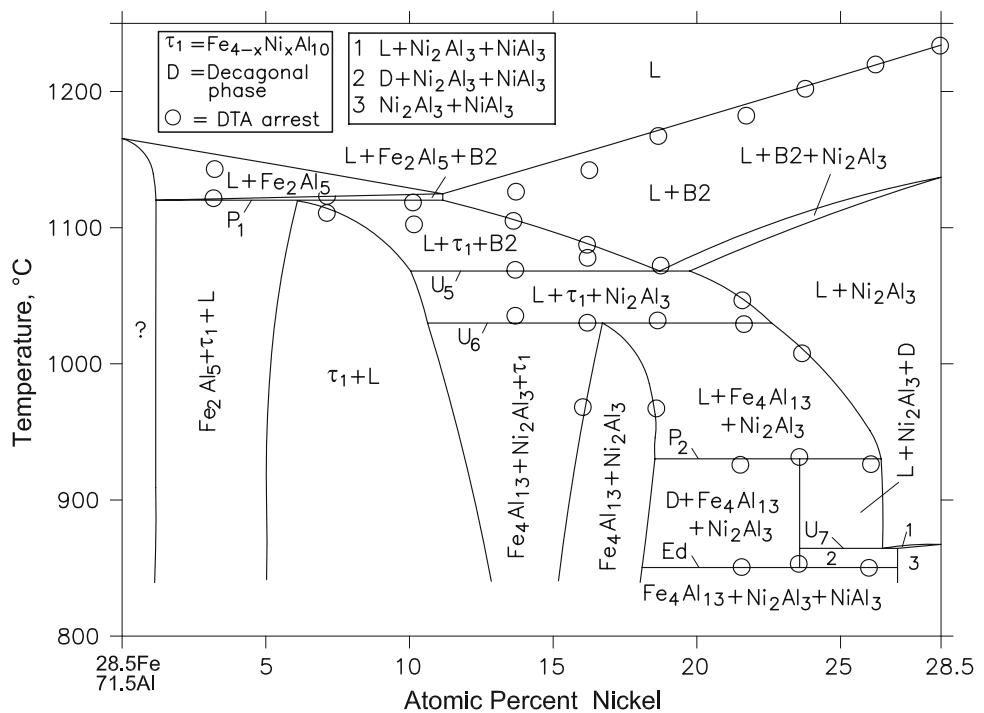


Fig. 2 Al-Fe-Ni vertical section at 71.5 at.% Al [2007Chu]

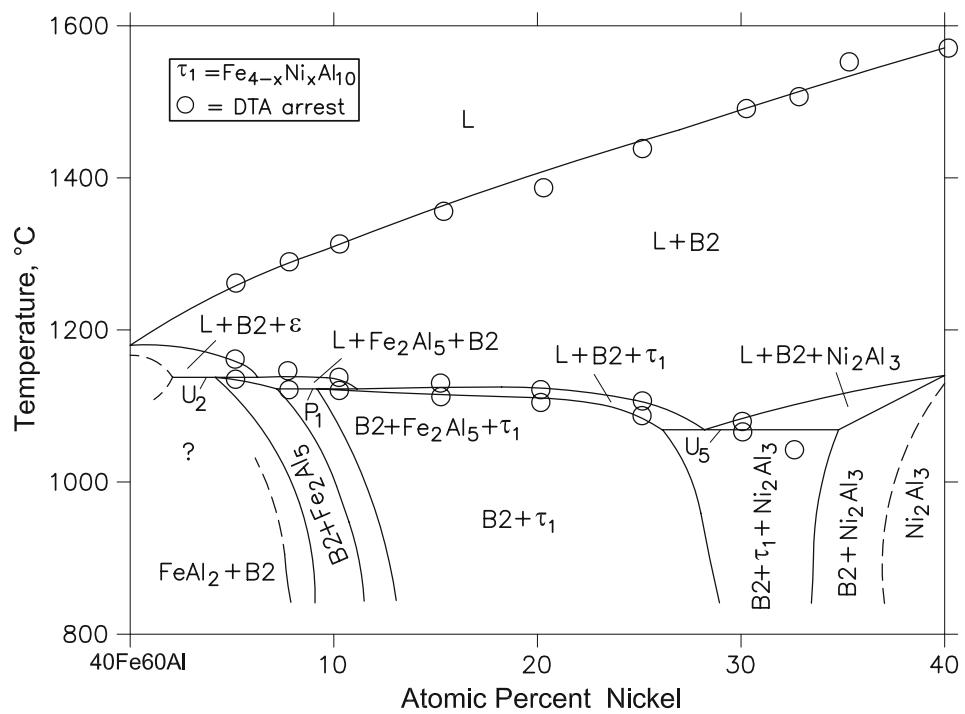
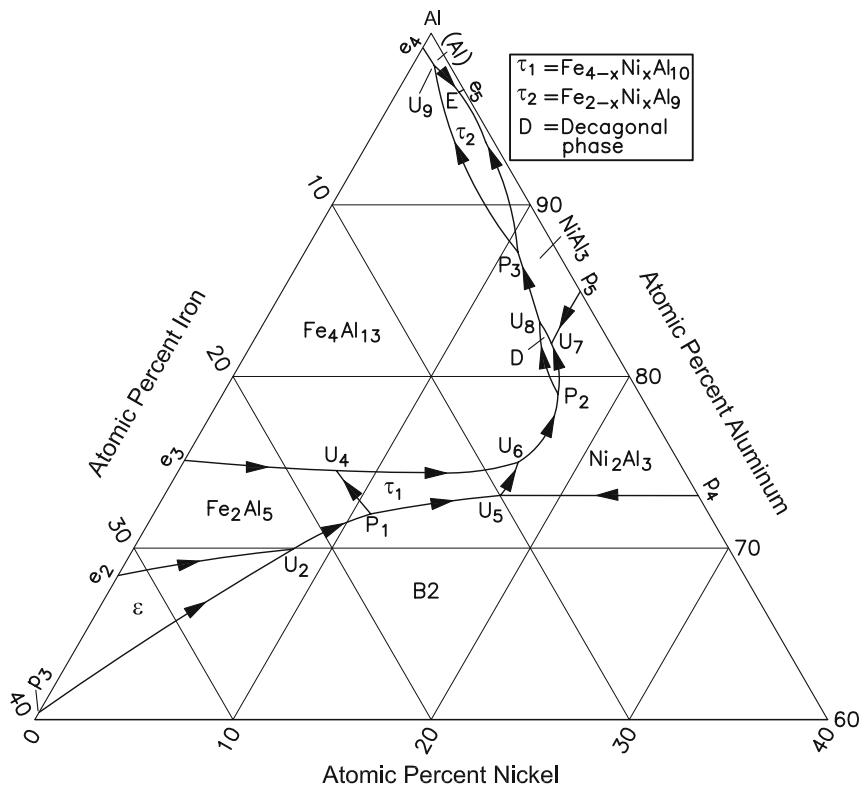


Fig. 3 Al-Fe-Ni vertical section at 60 at.% Al [2007Chu]

## Section II: Phase Diagram Evaluations



**Fig. 4** Al-Fe-Ni liquidus projection for Al-rich alloys [2007Chu]

$\delta$  and liquid that yields the Fe-based fcc solid solution. A continuous fcc solid solution denoted  $\gamma$  is stable over a wide range of temperature. At 517 °C, an ordered phase  $\text{FeNi}_3$  forms congruently from  $\gamma$ .

### Ternary Phase Equilibria

With starting metals of 99.999% Al, 99.99% Fe, and 99.99% Ni, [2007Chu] arc-melted under Ar atm 36 ternary compositions. The alloys were annealed at 850 °C for 2-3 weeks and quenched in water. The phase equilibria were studied with x-ray powder diffraction, electron probe microanalysis, and differential thermal analysis (DTA) at heating/cooling rates of 5 °C/min.

The isothermal section constructed by [2007Chu] at 850 °C is shown in Fig. 1. At this temperature, the ternary phase  $\text{Fe}_{4-x}\text{Ni}_x\text{Al}_{10}$  ( $\tau_1$ , denoted as  $\text{Fe}_3\text{NiAl}_{10}$  in earlier references) is stable. It has a homogeneity range of  $0.78 \leq x \leq 1.80$  and has the  $\text{Co}_2\text{Al}_5$ -type hexagonal structure with lattice parameters of  $a = 0.76781$ - $0.77049$  nm and  $c = 0.77131$ - $0.76432$  nm. The quasicrystalline decagonal phase D (denoted q by [2007Chu]) is stable at this temperature and has a fixed composition of  $\text{Fe}_{4.9}\text{Ni}_{23.4}\text{Al}_{71.7}$  [2007Chu]. The solubility of Fe and Ni in

$\text{Ni}_2\text{Al}_3$  and  $\text{Fe}_4\text{Al}_{11}$  is 4 and 11 at.%, respectively. The solubility of the third component in  $\text{Fe}_2\text{Al}_5$ ,  $\text{FeAl}_2$  and  $\text{NiAl}_3$  is small, between 1 and 2 at.%. Using the DTA results, [2007Chu] constructed five vertical sections at 80, 75, 71.5, 67 and 60 at.% Al, respectively. Figures 2 and 3 show the vertical sections at 71.5 and 60 at.% Al. The reactions at the invariant horizontals are marked. The decagonal phase D is stable between 930 and 850 °C, as seen in the section at 71.5 at.% (Fig. 2), whereas it is not present in the section at 60 at.% Al.

[2007Chu] combined their DTA results with the literature data to construct a complete liquidus surface for the Al-rich region. This is shown in Fig. 4. The ternary phase  $\text{Fe}_{2-x}\text{Ni}_x\text{Al}_9$  ( $\tau_2$ ,  $\text{Co}_2\text{Al}_9$ -type monoclinic) forms through the peritectic reaction  $P_3$  at 809 °C. A reaction sequence written by [2007Chu] is given in Fig. 5. For the sake of completion, the reactions in the Al-lean region and the solid-state reactions have been added. No distinction is made between bcc and the ordered forms of bcc. For the liquidus projection in the Al-lean region, see [1988Ray]. The sequential numbering of the reactions in Figs. 4 and 5 follows the usual direction of decreasing temperature and may not tally with that adopted by [2007Chu]. The reaction sequence is consistent with the triangulations in the isothermal section at 850 °C (Fig. 1) and the liquidus projection (Fig. 4).

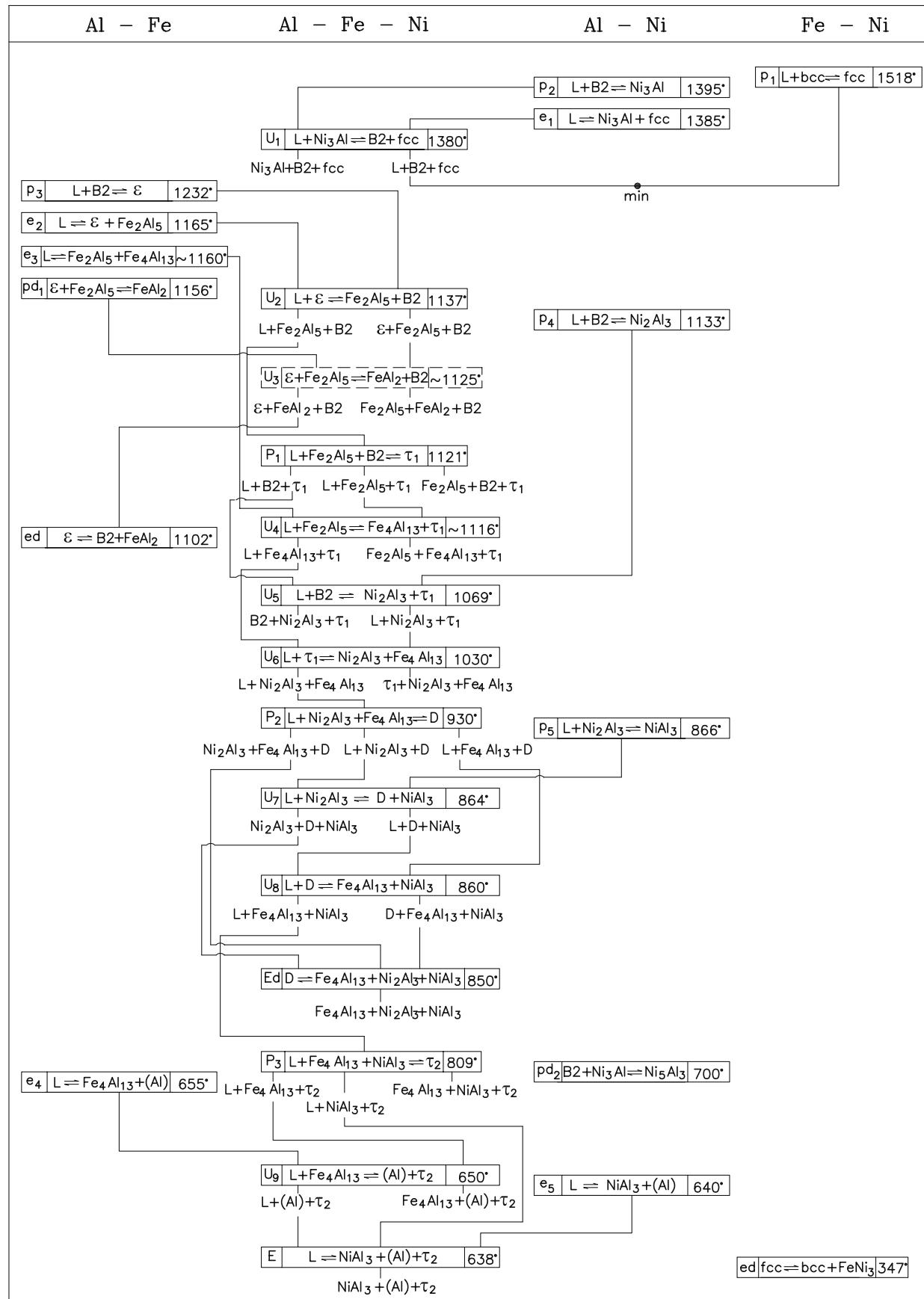


Fig. 5 Al-Fe-Ni reaction sequence during solidification [after 2007Chu]

## Section II: Phase Diagram Evaluations

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